

## Chemistry 12 Intext Solution Unit 2

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Question: 1.11 - Name the parameters that characterise a unit cell. Answer: Unit cells are characterized on six parameters – dimensions along three edges and three angles between their edges, i.e. a, b, c which are edges and  $\alpha$ ,  $\beta$  and  $\gamma$  which are angles between the edges. Question: 1.12 - Distinguish between (i) Hexagonal and monoclinic unit ...

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10.2. Why is sulphuric acid not used during the reaction of alcohols with KI? To solve the problem, H<sub>2</sub>SO<sub>4</sub> is replaced by phosphoric acid (H<sub>3</sub>PO<sub>4</sub>) which provides H<sup>+</sup> for the reaction and does not give I<sub>2</sub> as is done by H<sub>2</sub>SO<sub>4</sub>. 10.3. Write structures of different dihalogen derivatives of propane. 10.7 ...

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#### NCERT Solutions Class 12 Chemistry UNIT 1 THE SOLID STATE ...

Question 3.11: Suggest a list of metals that are extracted electrolytically. Question 3.12: Consider the reaction:  $\text{Cr}_2\text{O}_7^{2-} + 14\text{H}^+ + 6\text{e}^- \rightarrow 2\text{Cr}^{3+} + 8\text{H}_2\text{O}$  What is the quantity of electricity in coulombs needed to reduce 1 mol of  $\text{Cr}_2\text{O}_7^{2-}$ ? Question 3.14: Suggest two materials other than hydrogen that can be used as fuels in fuel cells.

#### NCERT Solutions Class 12 Chemistry UNIT 3 ELECTROCHEMISTRY ...

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Electrochemistry InText Solution. Question 1: How would you determine the standard electrode potential of the system  $\text{Mg}^{2+}|\text{Mg}$ ? Answer: We know that it is not possible to measure the electrode potential of a half cell. We can only measure the difference between electrode potentials of two half cells, which gives the cell potential of the cell.

#### Class 12 Chemistry Electrochemistry In Text Questions Solution

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