

## Chapter 6 Periodic Table Answers

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174 Chapter 6 • The Periodic Table and Periodic Law Section 6.1.1 Development of the Modern Periodic Table MAIN Idea The periodic table evolved over time as scientists discovered more useful ways to compare and organize the elements. Answers Elements and the 25. The mass of an electron is...

### Chapter 6 The Periodic Table Chapter Test A Answer Key:

Section 6.1 Development of the Modern Periodic Table 1. octaves 2. eight 3. nine 4. accepted 5. Dmitri Mendeleev 6. atomic mass 7. elements 8. atomic number 9. Henry Moseley 10. protons 11. periodic law 12. properties 13. 14.007 u 14. 7 15. [He]2s 22p 3 16. Nitrogen 17. N 18. b 19. c 20. d 21. a 22. e 23. three 24. metals 25. true 26. Group 2 27. true 28. noble gases 29. nonmetals Section 6.2 Classification of the Elements 1. h 2. f 3. n 4. a 5. k 6. d 7. l 8. o 9. i

### Ch 6 Study Guide answers

Chemistry Chapter 6 Study Guide. Elements in groups 3-12 that s contained in the d-block of the periodic table and, with some exceptions, is characterized by a filled outermost s orbital of energy level n, and filled or partially filled d orbitals of energy in level n-1.

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The Periodic Table: Chapter Problems Periodic Table Class Work 1. As you move from left to right across the periodic table, how does atomic number change? 2. What element is located in period 3, group 13? 3. What is atomic number of the element in period 6, group 2? 4. Will the element located at period 6, group 3 have a larger or smaller atomic

### The Periodic Table: Chapter Problems Periodic Table

Chapter 6 The Periodic Table 139 Key to periodic table A Key to periodic table B Thousands of periodic tables have been published since Mendeleev published his table. Each one is a little different from the rest. Shown above are keys to two of the hundreds of periodic tables now available. A key is an example or roadmap for using 3. 6. 163 63

### 05 CTR ch06 7/9/04 3:27 PM Page 131 ORGANIZING THE ELEMENTS 6

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PERIODIC TABLE: Dmitri Mendeleev -mid 1800's-proposed a table for 70 elements based on increasing mass and similar properties Henry Moseley -1913-determined the atomic number of elements and arranged the

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**[HOT!] Chapter 6 The Periodic Table Chapter Test B Answer ...**

Play this game to review undefined. The vertical (up and down) columns in the Periodic Table are called

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